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Note

High-performance liquid chromatographic determination of *tert*-butylhydroquinone in vegetable oils

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tert-Butylhydroquinone (TBHQ) is used as an anti-oxidant in vegetable oils. It can be used alone or in combination with other synthetic anti-oxidants¹. For the quantitative determination of TBHQ a spectrophotometric² or a gas chromatographic (GC) method³ can be applied. Although the GC method is more specific it is time- and labour-consuming as it is necessary to extract and derivatize the TBHQ. Recently, a gel permeation chromatographic method for the analysis of TBHQ and other anti-oxidants was reported⁴. Analysis time for this method is *ca.* 30 min and the detection limit for TBHQ is 500 ng. We report here a rapid and specific high-performance liquid chromatographic (HPLC) method for the quantitative determination of TBHQ in oils.

EXPERIMENTAL

Reagents

Dioxane containing 0.1 g water/100 g, *n*-hexane (spectroscopic grade) containing 0.004 g water/100 g and Oxinex 2378 were obtained from Merck (Darmstadt, G.F.R.). TBHQ was obtained from Eastman (Kingsport, Tenn., U.S.A.), butylated hydroxytoluene (BHT), butylated hydroxyanisole (BHA) and 2,4,5-trihydroxybutyrophene (TBP) from BDH (Poole, Great Britain) and dodecyl gallate (DG) from Naarden (Wormerveer, The Netherlands).

Equipment

The HPLC equipment consisted of a Varian 5000 pump, Valco inlet valve and a Farrand Mk 1 spectrofluorometer, equipped with a 7- μ l flow cell. A primary wavelength of 309 nm, a secondary wavelength of 340 nm and a slit width of 10 nm were used. Peak retention times and areas were determined with a Hewlett-Packard 3352B laboratory data system. The stainless steel column (25 \times 0.4 cm I.D.) was slurry-packed⁵ with 5 μ m LiChrosorb SI 60 (Merck).

Method

The mobile phase consisted of dioxane and *n*-hexane (24:76, v/v), pumped at a flow-rate of 3 ml/min. TBHQ standard solutions containing 1, 2, 2.8, 4 and 10

mg/100 ml in *n*-hexane were prepared. A calibration curve (peak areas against concentration) was obtained by injecting 13.9 μ l of each standard solution in duplicate. TBHQ in oil samples was determined by direct injection of 13.9 μ l oil. Recovery determinations were conducted by adding 20 mg TBHQ to 100 g maize oil (oil density 0.918 g/cm³ at 20°).

RESULTS AND DISCUSSION

A typical chromatogram of TBHQ dissolved in oil is reproduced in Fig. 1. TBHQ eluted with a capacity factor of $k' = 2.1$ from a column giving 4600 theoretical plates. The maize oil tocopherols, which are also detectable under the prescribed conditions of analysis, eluted with a capacity factor close to zero. An oil sample without TBHQ did not show any peaks at or near $k' = 2.1$.

Solutions of Oxinex 2378, BHT, BHA, DG and TBP in *n*-hexane were prepared and injected. Only BHA gave a detectable peak which eluted with a k' value similar to that of the tocopherols. Subsequently a mobile phase of 4% dioxane in

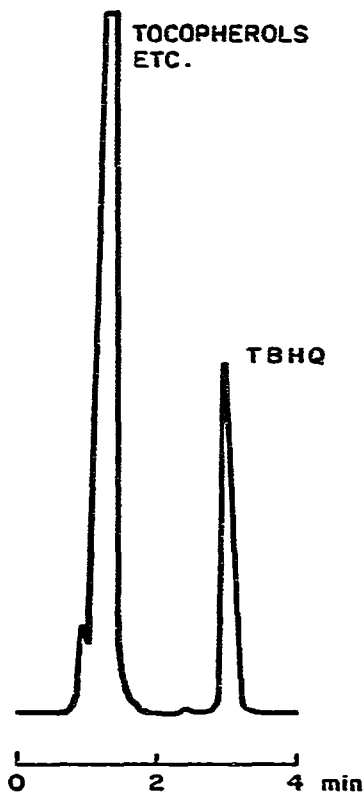


Fig. 1. Chromatogram of 20 mg TBHQ/100 g maize oil. Steel column (25 × 0.4 cm I.D.) slurry-packed with LiChrosorb SI 60 (5 μ m), with dioxane and *n*-hexane (24:76) as mobile phase (3 ml/min) and a fluorometric detector.

n-hexane, which was more suitable for the separation of the tocopherols⁶, was used and we found that BHA eluted with γ -tocotrienol.

A linear calibration curve for TBHQ was obtained and the least-squares method was applied to calculate the line of best fit for the data ($y = 48.3x - 2.2$, $r = 0.9998$). A recovery figure of 98.5% for TBHQ from maize oil was obtained with a coefficient of variation of 2.4% for eight determinations. The detection limit was 6 ng, and the average analysis time per sample was 5 min. The described method is therefore fast, sensitive and convenient.

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